

MATERIAL SAFETY DATA SHEET

ANHYDROUS AMMONIA

DISTRIBUTOR:

W.D. SERVICE COMPANY, INC.

780 CREEK ROAD, P.O. BOX 147

BELLMAR, NJ 08099

CORPORATE EMERGENCY TELEPHONE NUMBER: 856-931-6100, 24 HOUR: 800-424-9300

DESCRIPTION

CHEMICAL NAME: Ammonia, Anhydrous

SYNONYMS: Ammonia, Liquefied.

CHEMICAL FAMILY: Ammonia

COMPOSITION: 99+% Ammonia.

CAS REGISTRY NO.7664-41-7

FORMULA: NH₃

MOL. WT. 17.03(NH₃)

STATEMENT OF HEALTH HAZARD

HAZARD DESCRIPTION: Irritant and corrosive to skin, eye, respiratory tract and mucous membranes.

May cause severe burns, eye and lung injuries. Skin and respiratory related diseases aggravated by exposure. Not recognized by OSHA as a carcinogen. Not listed in the National Toxicology Program annual report. Not listed as a carcinogen by the International Agency for Research on Cancer.

EXPOSURE LIMITS:

50 ppm PEL - Federal OSHA

35 ppm STEL - NIOSH

25 ppm TWA - NIOSH

300 ppm IDLH - NIOSH

EMERGENCY TREATMENT

EFFECTS OF OVEREXPOSURE: Eye: lachrymation, edema, blindness. Skin: irritation, corrosive burns, blister formation. Contact with liquid will freeze the tissue, then produces a caustic burn. Inhalation: heavy, acute exposure may result in severe irritation of the respiratory tract, glottal edema, bronchospasm, pulmonary edema and respiratory arrest. Chronic effects: bronchitis. Extreme exposure (5000 ppm) can cause immediate death from spasm, inflammation or edema of larynx.

EMERGENCY AID: Skin: flush with copious amounts of water while removing contaminated clothing and shoes. Do not rub, or apply ointment on affected area. SEEK IMMEDIATE MEDICAL HELP. Ingestion: if conscious, give large amount of water to drink. Refer immediately to physician. Eye: flush with copious amounts of water for 15 min. Eyelids should be held apart and away from eyeball for thorough rinsing. SPEED AND THOROUGHNESS IN RINSING THE EYE IS MOST IMPORTANT IN PREVENTING LATENT PERMANENT INJURIES. Inhalation: remove to fresh air. Administer oxygen or artificial respiration if necessary.

NOTE TO PHYSICIAN: Lung injury may appear as delayed phenomenon, pulmonary edema may follow chemical bronchitis. Supportive treatment with necessary ventilation actions, including oxygen, may warrant consideration.

PHYSICAL DATA

BOILING PT: -33°C (-28°F)	FREEZING PT: -78°C (-108°F)
VAPOR DENSITY (AIR=1): 0.596 @ 0°C (32°F)	VAPOR DENSITY (Air=1): 0.6
VAPOR PRESSURE: 10 atm @ 25.7°C	SOLUBILITY IN WATER: 89.9 g/100cc @ 0°C 7.4 g/100cc @100°C
SPECIFIC GRAVITY (H ₂ O=1): 0.682 @ 4°C (39°F)	EVAPORATION RATE (Water=1) Faster than water
PERCENT VOLATILE: 100%	SURFACE TENSION: 23.4 Dynes/cm @ 11.1°C
APPEARANCE & ODOR: Colorless gas/liquid and pungent odor	

FIRE AND EXPLOSION HAZARD DATA

FLASH POINT:	None
AUTOIGNITION TEMP:	651 °C (1204°F) catalyzed by iron; 850 °C (1562°F) uncatalyzed
FLAMMABLE LIMITS IN AIR:	LEL 15% UEL 28%
EXTINGUISHING MEDIA:	Non-combustible

SPECIAL FIRE-FIGHTING PROCEDURES: Must wear protective clothing and respiratory protection. See PROTECTIVE EQUIPMENT. Stop source if possible. Cool fire exposed containers with water spray. Stay upwind and use water spray to knock down vapor and dilute.

UNUSUAL FIRE AND EXPLOSION HAZARDS: Not generally a fire hazard. If relief valves are inoperative, heat-exposed storage containers may become explosion hazards. Ammonia contact with chemicals such as mercury, chlorine, iodine, bromine, silver oxide, or hypochlorites can form explosive compounds. Special hazards with chlorine to form chloramine gas, also a primary skin irritant and sensitizer. Combustion may form toxic nitrogen oxides.

CHEMICAL REACTIVITY

STABILITY: Stable at room temperature. Ammonia will react exothermically with acids and water.

CONDITIONS TO AVOID: Avoid mixing with sulfuric acid or other strong mineral acids. Avoid mixing with hypochlorites (chlorine bleach) or other halogens and sodium hydroxide. Avoid contact with galvanized surfaces, copper, brass, bronze, aluminum alloys, mercury, gold, silver, and strong oxidizers. Avoid heating.

HAZARDOUS DECOMPOSITION PRODUCTS: Hydrogen and nitrogen gases above 450°C (842°F)

SPILL OR LEAK PROCEDURES

STEPS TO BE TAKEN: Wear respiratory protection and protective clothing, see PROTECTIVE EQUIPMENT. Stop source if possible. If Exposure concerns are present, stay upwind and use water spray downwind of leak source to absorb ammonia gas and dilute. CAUTION: ADDING WATER DIRECTLY TO LIQUID SPILLS WILL INCREASE VOLATILIZATION OF AMMONIA, THUS INCREASING POSSIBILITY OF EXPOSURE.

WASTE DISPOSAL: Listed as hazardous substance under CWA (40 CFR 116.40 CFR 117.3 Reportable Quantity 100 lbs. OR 45.4kg) Classed as a hazardous waste under RCRA (40 CFR 261.32 Corrosive # D002). Comply with all regulations. Suitably diluted product may be disposed of on agricultural land as fertilizer. Keep spill from entering streams or lakes.

SPECIAL PROTECTION AND PROCEDURES

RESPIRATORY PROTECTION: MSHA/NIOSH approved respiratory protection with full face piece for gas and vapor contaminants effective for anhydrous ammonia and able to be used for entry and escape in emergencies. Refer to 29 CFR 1910.134 and ANSI: Z88.2 for requirements and selection.

VENTILATION: Local exhaust sufficient to keep ammonia gas below Permissible Exposure Limits. Refer to 29 CFR 1910.134 and ANSI: Z9.2 for requirements and selection.

PROTECTIVE EQUIPMENT: Splash-proof, chemical safety goggles, rubber gloves and boots to prevent contact. Respiratory protection. Cotton work clothes recommended. Refer to 29 CFR 1910.132 to 1910.136 for requirements.

SPECIAL PRECAUTIONS

STORAGE AND HANDLING: Store in cool, well-ventilated area with containers tightly closed. OSHA 29 CFR 1910.111 prescribes handling and storage requirements for anhydrous ammonia as a hazardous material.

WORK-PLACE PROTECTIVE EQUIPMENT: as discussed above should be near, but outside of ammonia area. Eyewash and safety shower in immediate vicinity. See 29 CFR 1910.141 for workplace requirements.

DISPOSAL: Ammonia is listed as a hazardous substance under FWPCA. See WASTE DISPOSAL. Classified as RCRA Hazardous waste due to corrosivity.

PERSONAL: Avoid unnecessary exposure. Use protective equipment as needed. Do not wear contact lenses.

LABELING AND SHIPPING

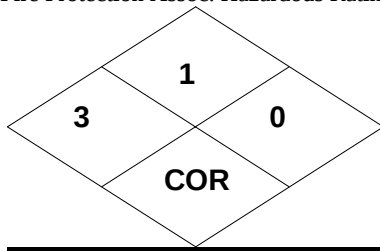
HAZARD CLASS: 2.2 (Nonflammable Gas) U.S. Domestic AND 2.3 (Poison Gas) International

PROPER SHIPPING DESCRIPTION: Ammonia, Anhydrous, 2.2, UN1005, RQ, Inhalation Hazard (U.S. Domestic) AND Ammonia, Anhydrous, 2.3, UN1005, RQ, Poison-Inhalation Hazard Zone "D" (International)

PLACARD/LABEL: Nonflammable Gas (U.S. Domestic) AND Poison Gas, Corrosive (Subsidiary) (International)

IDENTIFICATION NO: UN 1005

National Fire Protection Assoc. Hazardous Rating:



Hazardous Materials Identification System Labels:

ANHYDROUS AMMONIA	
HEALTH	3
FLAMMABILITY	1
REACTIVITY	0
PERSONAL PROTECTION	H

OTHER REGULATORY REQUIREMENTS

Under the Comprehensive Environmental Response, Compensation, and Liability Act of 1980 (CERCLA), Section 103, any environmental release of this chemical equal to or over the reportable quantity of 100 lbs. must be reported promptly to the National Response Center, Washington, D.C. (1-800-424-8802). Any consumer product containing 5% or more ammonia requires a POISON label under FHSA (16 CFR 1500. 129(1)).

The material is subject to the reporting requirements of Section 313, Section 304, Section 312, Title III of the Superfund Amendments and Reauthorization Act of 1986 and 40 CFR 372.

EPCRA extremely hazardous substance, 40 CFR 355, Title III, Section 302 – Ammonia, TPQ 500 lbs.

Regulated under Clean Air Act 40CFR112(r) , TQ 10,000 lbs.

EPA Hazard Categories - Immediate: Yes; Delayed: No; Fire: No; Sudden Release: Yes; Reactive: No

The information, data, and recommendations in this material safety data sheet relate only to the specific material designated herein and do not relate to use in combination with any other material or in any process. The information, data, and recommendations set forth herein are believed by us to be accurate. We make no warranties, either expressed or implied, with respect thereto and assume no liability in connection with any use of such information, data, and recommendations.

Revised: July, 2005